Magnetism is one of many basic physical properties of molecules which we chemists could explore and understand by studying the geometrical and electronic structures. This area has flourished as a lively crossroad for physics, chemistry and materials science. Single-molecule magnets (SMMs) are one of the most cutting-edge fields in molecular magnetism thriving from the inception of the noted [Mn12] in 1993. SMMs belong to a class of molecules which can exhibit slow magnetic relaxation purely of a molecular origin rather than the collective behavior facilitated by magnetic exchange coupling in ensembles through a 3D network. If there’s only a single spin center, they’re also noted as single-ion magnets (SIMs). The famed [Pc2Tb] – [{Pc}2] – dianion of phthalocyanine opened the prosperous stage to lanthanide SIMs (Ln-SIMs) because of their unparalleled single-ion anisotropy due to unquenched orbital angular momentum especially for heavy lanthanide ions. The success of [Pc2Tb] – [{Pc}2] – offers an intriguing possibility for molecular spintronics. In 2011, an organometallic SIM (Cp*Er(COT)) was reported by our group. It shows two magnetic relaxation processes under a zero applied DC magnetic field arising from two statically disordered conformers with energy barriers of 197 and 323 K, respectively, which were several times higher than those of cluster-based SMMs at that time. From then on concerted efforts have been made toward Ln-SIMs with a high energy barrier (∆/k_B) for the reversal of magnetic quantum states and magnetic blocking temperature (T_B).*

In order to construct a good SIM, hard bases containing O or N coordinating atoms are often introduced to enhance the uniaxial ligand field and hence the zero field splitting which is the main source of Ising-type anisotropy for Ln-SIMs. But in this work we do exactly the opposite. Based on (Cp*)Er(COT), using a soft phosphorus analogue of the Cp− ligand – a phosphacyclopentadienyl (phospholyl) ligand – can improve the SIM performance unexpectedly. Phospholyl ligands have already become well-established alternatives to cyclopentadienyl (Cp−) groups in organometallic chemistry for polymerization catalysis and have captivated synthetic chemists in coordination chemistry. Thanks to the isolobal analogy between −C(=H)Me and phosphorus, phospholyl ligands can retain aromaticity to some extent relative to Cp− or [Cp*]−. And they still tend to keep their η5 mode like Cp− in coordinating with Ln ions in most cases. However, according to the theory of hard and soft acids and bases, the substitution of C by a larger and floppy P makes them softer bases and poor π-donors. This substitution will undoubtedly lead to the lowered affinity of phospholyl ligands with some high-valent metal ions Ln(III) for instance which act as hard acids. Furthermore, in Dsp− two electron-withdrawing groups −SiMe3 further reduce the nucleophiliicity. Meanwhile, they are less inclined to be in η5 coordination mode using the lone-pair electrons of P atoms for the lower affinity of P with Ln ions with only three exceptions. So we can safely replace the Cp− ring by a phospholyl group and keep the same coordination...
mode, and make a rational structural comparison between the two. For Ln(II), the lower oxidation state, larger ionic radius and hence softer acidity make it more easily bind to phospholyl ligands. Phospholyl Ln(II) complexes can be synthesized either by salt metathesis or by a redox reaction between metal Ln and 1,1′-biphosphoryls. However, especially for heavy Ln(III), it’s a different scenario. Comparatively very few phospholyl heavy Ln(III) complexes have been successfully synthesized and structurally characterized so far, such as, (Dtp)2TmI (Dtp = 2,2,6,6-tetramethyl-3,5-heptadiyne-2,4-diyldipropadiyl), (Dsp)Nd(COT)(THF), (Dtp)2DyI, [(Dsp)2DyI]2 (ref. 23) and [(Dtp)2Tm(m-Si)]2. Moreover, the single crystal structure of sandwiched (phospholyl)Ln(COT) has not been reported yet, and only two close examples of early Ln(III) compounds will be different accordingly, which may bring about significant variation of optical and magnetic properties. Alternating current (AC) magnetic susceptibility measurement of (Dsp)Er(COT) demonstrated a significant change of magnetic dynamics compared with its parent analogue (Cp*)Er(COT) (vide infra).

Following the work by François Nief and Louis Ricard with some modifications, we first get phospholyl potassium K(Dsp), which then reacts with ErI(COT)(THF), (x = 2–3, see the ESIF) in toluene under reflux to produce (Dsp)Er(COT) with insoluble KI (Scheme 1). Trying to precipitate KCl by reacting with (THF)Er(COT)(μ-Cl)]2 is unsuccessful, although K(Dsp) can eliminate Cl− in {(THF)2Nd(COT)(μ-Cl)]2 (ref. 25) which is probably due to different acidity between light and heavy Ln(m) ions (vide supra). Orange single crystals suitable for X-ray diffraction analysis can be obtained by cooling the concentrated toluene solution of (Dsp)Er(COT) at ~30 °C for several days (anal. calc.: C, 45.59%; H, 6.12%; found: C, 45.73%; H, 6.12%).

As depicted in Fig. 1, the single crystal structure of (Dsp)Er(COT) at 180 K (structure refined according to ref. 28) reveals itself as a double-decker structure, reminiscent of its Cp* parent. It crystallizes in the P21/c space group. The nearest Er···Er distance is 7.832 Å. It’s noteworthy that the COT2− ring is in static disorder as well with the ratios 0.56 and 0.44 (Fig. S4 in the ESI†). The tilting angle between Dsp− and COT2− planes is 10.5°. The distance from Er(m) to the five-membered ring is 2.321 Å, larger than that in (Cp*)Er(COT) (2.268 Å), which crystallographically shows a weaker coordination between the two, yet Er(m) is much closer to COT2− (1.686 Å in [Dsp]Er(COT), 1.727 Å in the Cp* version and 1.875 Å in [Er(COT)2−]). The Er−C(Dsp−) bond lengths range from 2.635 to 2.645 Å, while Er−C(COT2−) bond lengths are from 2.471 to 2.491 Å. The Er−P bond length is 2.793 Å, being close to the reported Er−P bond length.11 To the best of our knowledge, this is the first sandwich structure reported with a phospholyl ligand and COT2− coordinating simultaneously to a heavy Ln(m).

We wonder if a slight change of only one coordinated atom could trigger a pronounced difference of magnetic properties. We resort to AC susceptometry which is often applied to probe magnetic dynamics. The out-of-phase component (χ_m') of molar AC susceptibility for (Dsp)Er(COT) exhibits noticeable

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**Scheme 1** Synthesis of (Dsp)Er(COT).

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**Fig. 1** Front view (a), top view (b) and side view (c) of the single crystal structure of (Dsp)Er(COT) with pink, orange, tan, and black ellipsoids (50% possibility) representing Er, P, Si, and C, respectively (another disorder fragment of COT2− is shown in Fig. S4†). Hydrogen atoms and another disordered part of COT2− (Fig. S4†) have been omitted for clarity. Selected bond lengths: Er−P = 2.793 Å; Er−C(Dsp−) = 2.635−2.645 Å; Er−C(COT2−) = 2.471−2.510 Å.
frequency-dependence over 1 to 1000 Hz from 2 to 28 K under a zero applied DC field (Fig. 2). As with its prototype, this slow magnetic relaxation of (Dsp)Er(COT) unveils itself as a SIM. There is a single peak in the plot of $\chi''_m$ versus $T$ at a given frequency while there are two for (Cp*)Er(COT). The narrow distribution of relaxation time $\tau (\alpha = 0.01–0.2)$ from the fitting of the Argand plot26 (Fig. S10†) indicates almost a single relaxation despite the existence of two static disorder conformers which instead gives rise to two relaxation processes corresponding to two peaks of $\chi''_m$ in the measured temperature range for (Cp*)Er(COT)1.

A combination of an Orbach process and quantum tunneling of magnetization (QTM) ($\tau^{-1} = \tau_0^{-1} \exp[-\Delta/(k_B T)] + \tau_{\text{QTM}}^{-1}$, $\tau$ is the relaxation time) can fit the data of $\ln \tau$ versus $T$ well (Fig. 3). It demonstrates that the thermally activated Orbach process is dominant above 12 K and QTM considerably masks other possible relaxations below 10 K. The fitting returns a relaxation barrier $\Delta/k_B = 358(3) \text{ K}$ which is much higher than that of (Cp*)Er(COT) (197, 323 K)9 and [Er(COT)$_2$] (206 K), and among the highest ones in Er(III)-based SIMs.30,33 Magnetic dynamics measurements were also performed under a 1000 Oe DC field to effectively suppress the QTM (Fig. S12†). A similar workup to that described above gives $\Delta/k_B = 367(1) \text{ K}$ (Fig. S13†).

The subsequent measurement of zero field cooled and field cooled magnetization (ZFC-FC) presents a divergence at about 8.5 K between the two curves (Fig. S7†), which reminds us of a probable magnetic hysteresis. As expected, typical butterfly-shaped magnetic hysteresis loops in mesoscopic SMMs8,9 can be observed below 9 K with a field-sweeping rate of 200 Oe s$^{-1}$ (Fig. 4 and S8†). At 2 K, there is still a magnetic remnant of 0.3 Nβ at a zero field and the value of the coercive field $H_c$ is 55 Oe. The

![Fig. 2](image1.png)

**Fig. 2** Temperature (a) and frequency-dependence (b) of molar AC susceptibility (upper: $\chi''_m$, lower: $\chi'_m$) for (Dsp)Er(COT) under a zero applied DC field over the temperature range of 2–28 K and the frequency range of 1–1000 Hz.

![Fig. 3](image2.png)

**Fig. 3** Plot of natural log of relaxation time versus inverse temperature under a zero applied DC field. The purple and red circles represent the data from MPMS and PPMS, respectively. The black solid line represents the best fitting using a combination of Orbach and QTM processes. The fitting gives $\Delta/k_B = 358(3) \text{ K}$ and $\tau_0 = 1.6(3) \times 10^{-11} \text{ s}$.

![Fig. 4](image3.png)

**Fig. 4** Variable-field magnetization for (Dsp)Er(COT) with a field-sweeping rate of 200 Oe s$^{-1}$. (Inset) Expanded view of the variable-field magnetization near the zero field at 9 K and 10 K.
magnetic blocking temperature \(T_B\) of about 9 K (Fig. S8†) is much higher than that of \([\text{Cp}^*]_{\text{Er(COT)}}\) (5 K, 550 Oe min\(^{-1}\)) and almost rivals that of \([\text{Er(COT)}_2]^{-}\) (10 K, 0.78 mT s\(^{-1}\)).\(^{28}\) This is a robust confirmation that \([\text{Dsp}]_{\text{Er(COT)}}\) is a magnet which can reserve its magnetization even under a zero field like a block magnet. The above results explicitly unveil \([\text{Dsp}]_{\text{Er(COT)}}\) as a SIM with a good performance.

Why does the seemingly weaker ligand field of Dsp\(^{-}\) produce stronger anisotropy than \([\text{Cp}^*]_{\text{Er(COT)}}\)? For one thing, we think that the phospholyl ligand and its parents \(\text{Cp}^*\) and \([\text{Cp}^*]_2\) are more like axial ligands favorable for the oblate electron distribution of Dy\((m)\) to stabilize its Ising-type \(M\) ground state\(^{24,35}\) as exemplified by plenty of excellent \(\text{Cp}\)-based Dy\((m)\) SMMs.\(^{26}\) But they're unsuitable for prolate Er\((m)\) under Ising-limit conditions. This sort of difference between Dy\((m)\) and Er\((m)\) originates from the opposite signs of axial crystal field parameters \(B^0\) and \(B^2\)\(^{ab}\) for the two lanthanide ions. The moving away of the Dsp ligand from the axial electron cloud of Er\((m)\) could reduce the electron repulsion. On the other hand, a large equatorial ligand such as the COT\(^2\) ring with high rotational symmetry is suitable for prolate Er\((m)\) to stabilize its \(M\) ground state as corroborated by \([\text{Cp}^*]_{\text{Er(COT)}}\) and \([\text{Er(COT)}_2]^{-}\). The closer the Er\((m)\) is to the COT\(^2\) ring, the stronger the ligand field the \(-2\) charged COT\(^2\) exerts on Er\((m)\), and hence the larger the anisotropic splitting of the ground \(J\) state. Although there are two COT\(^2\) rings in \([\text{Er(COT)}_2]^{-}\),\(^{36}\) the distance between Er\((m)\) and COT\(^2\) (\(d_{\text{Er-COT}}\)) becomes larger (1.875 Å) due probably to the repulsion between the two COT\(^2\) rings which results in a weakened ligand field instead and its anisotropic energy barrier is lower than that of \([\text{Cp}^*]_{\text{Er(COT)}}\). In this regard, the closer the Er\((m)\) is to the COT\(^2\) ring within a certain range, the lower and less mixed the ground state doublets are, making them closer to the Ising limit at least for the lowest Kramers doublets. Compared with \([\text{Cp}^*]_{\text{Er(COT)}}\), the weak adhesion of Er\((m)\) in \([\text{Dsp}]_{\text{Er(COT)}}\) with the phospholyl ligand leads to a shorter distance between Er\((m)\) and COT\(^2\) (vide supra) in order to maintain a balanced and conservative electrostatic potential of Er\((m)\). So SIM performance improves as elaborated above in spite of lower symmetry (Table 1). We believe that \(d_{\text{Er-COT}}\) plays an essential role in causing different magnetic dynamics between \([\text{Dsp}]_{\text{Er(COT)}}\) and \([\text{Cp}^*]_{\text{Er(COT)}}\).

As a statistical induction, we have found the COT\(^2\) Er\((m)\) SIMs reported to-date and plotted each scattered point representing the data of \(\Delta/k_B\) versus \(d_{\text{Er-COT}}\) (Fig. 5a). In a panoramic view, when the COT\(^2\) ring moves gradually away from Er\((m)\), the barrier goes through a precipitous fall below 1.7 Å, and then changes less with increasing \(d_{\text{Er-COT}}\). In order to theoretically understand the trend, we calculated the energy barrier of the model molecular fragment \([\text{Er(COT)}]^{-}\) with a different \(d_{\text{Er-COT}}\) using the \textit{ab initio} method\(^{37}\) (Table S5; Fig. 5b and S24†). Previous computation\(^{38,39}\) involved two COT\(^2\) rings based on the single crystal structure of \([\text{Er(COT)}_2]^{-}\). However, we think it’s not a single factor story in which high negative charge repulsion between COT\(^2\) rings may result in an unstable ground state within a certain distance (vide supra). So the result may not be very convincing if we only want to examine the relationship between \(\Delta/k_B\) and \(d_{\text{Er-COT}}\). In our model with only one COT\(^2\) ring, as expected, when \(d_{\text{Er-COT}}\) becomes larger than 1.6 Å, the energy barrier goes down approximately in a linear way. The downtrend of the energy barrier with increasing \(d_{\text{Er-COT}}\) in Fig. 5 shows that a short \(d_{\text{Er-COT}}\) is conducive to a high energy barrier, and the chemically adjustable soft P atom in the five-membered ring is the pushing hand.

<table>
<thead>
<tr>
<th>Table 1</th>
<th>Comparison of structural metrics and energy barrier between ([\text{Cp}^*]<em>{\text{Er(COT)}}) and ([\text{Dsp}]</em>{\text{Er(COT)}})</th>
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<tbody>
<tr>
<td>([\text{Cp}^*]_{\text{Er(COT)}})</td>
<td>([\text{Dsp}]_{\text{Er(COT)}})</td>
</tr>
<tr>
<td>Er-centroid (COT(^2))</td>
<td>1.7267(3)</td>
</tr>
<tr>
<td>Er-centroid (five-membered ring)</td>
<td>2.2679(3)</td>
</tr>
<tr>
<td>Tilting angle between the two rings</td>
<td>7.3</td>
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<td>(\Delta K)</td>
<td>197, 323</td>
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Fig. 5 (a) Scattered points representing the data of \(\Delta/k_B\) vs. \(d_{\text{Er-COT}}\) from reported COT-based Er\((m)\) SIMs.\(^{1,2,5,5,55,58}\) The data of \([\text{Cp}^*]_{\text{Er(COT)}}\) are from the calculation based on the ligand field for easy comparison. ([Tp]\(^{-}\) = tris(1-pyrazolyl)borate and [Tp\(^*\)]\(^{-}\) = tris(3,5-dimethyl-1-pyrazolyl)borate); (b) plot of energy barrier vs. series of \(d_{\text{Er-COT}}\) for the hypothetical fragment \([\text{Er(COT)}]^{-}\) calculated based on \textit{ab initio} inset: \([\text{Er(COT)}]^{-}\).
From the above elaboration, we can see the main line clearly whereby the introduction of a soft P atom leads to a larger distance from Er(III) to the Cp analogue which reflects a weaker affinity between the two, and hence a shorter $a_{\text{Er-COT}}$ which enhances the ligand field and anisotropic splitting. So a higher energy barrier is possibly reached.

In this series, we have also examined (Dsp)Ln(COT) (Ln = Tb (CCDC No. 1835954), Dy (CCDC No. 1835957), and Tm (CCDC No. 1835958)) among which (Dsp)Dy(COT) exhibits slow magnetic relaxation under a 2000 Oe applied DC field with $\Delta/k_B = 57$ K (Fig. S15†). (Dsp)Tm(COT) is also a field-induced SIM under a 2500 Oe applied DC field with an energy barrier of 109 K (Fig. S18†), which is almost the highest in Tm(III)-based SIMs so far.

Conclusions

We have successfully synthesized [Dsp]Er(COT) whose sandwiched single crystal structure is reported here for the first time. Compared with its parent compound (Cp*)Er(COT), the replacement of only one coordinated carbon atom on the cyclopentadienyl ligand by a softer phosphorus atom results in a considerable change of magnetic dynamics which returns a relaxation energy barrier of 358 K under a zero applied DC field and blocking temperature of 9 K. Both manifest it as a better SIM than (Cp*)Er(COT). The essential reason mainly resides in the short distance between Er(III) and COT$^{2-}$ caused by the weak affinity of Er(III) with Dsp. Through the introduction of a softer coordinating atom, we have obtained a better Er(III) SIMD. Inspired by this work, we will continue to focus on introducing more soft P atoms and expect better Er(III)-based SIMs.

Conflicts of interest

There are no conflicts to declare.

Acknowledgements

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Notes and references


