A mass spectrometry-based approach for identifying novel DNA polymerase substrates from a pool of dNTP analogues

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ABSTRACT

There has been a long-standing interest in the discovery of unnatural nucleotides that can be incorporated into DNA by polymerases. However, it is difficult to predict which nucleotide analogs will prove to have biological relevance. Therefore, we have developed a new screening method to identify novel substrates for DNA polymerases. This technique uses the polymerase itself to select a dNTP from a pool of potential substrates via incorporation onto a short oligonucleotide. The unnatural nucleotide(s) is then identified by high-resolution mass spectrometry. By using a DNA polymerase as a selection tool, only the biologically relevant members of a small nucleotide library can be quickly determined. We have demonstrated that this method can be used to discover unnatural base pairs in DNA with a detection threshold of ≤10% incorporation.

INTRODUCTION

DNA polymerases are ubiquitous and necessary enzymes for all living organisms. During DNA replication, polymerases normally only use four different nucleotides; however, a great number of other nucleotides are able to act as substrates for various polymerases. Many of these nucleotides or nucleotide analogs have found therapeutic applications, whereas others are useful tools for biochemical and biological studies. It is, therefore, of great interest to develop new methodology to identify novel nucleotides that can act as substrates for DNA polymerases.

All of the major replicative polymerases share a similar overall conformation and appear to use a similar two metal catalytic center, implying a similar mechanism of nucleotide incorporation (1,2). However, the mechanism(s) by which different polymerases select between right and wrong (d)NTPs vary substantially, as can be seen by the differential incorporation of various analogs by different polymerases (3–6). For example, DNA pol α, a B family polymerase, does not require the base pair formed between the incoming dNTP and the template base being replicated to have either a similar shape as a correct base pair or Watson–Crick hydrogen bonds. In contrast, both human and herpes primase choose whether or not to polymerize a NTP based on the formation of Watson–Crick hydrogen bonds between the incoming NTP and the template base.

A further complication in nucleotide design is the differential interaction of polymerases with a base analog present in a dNTP versus in the template. Although a given analog may act as an acceptable substrate when in the template strand, it may not be incorporated as a dNTP, or vice versa (3,7,8). In combination with the different mechanisms of nucleotide selectivity, this lack of consistency in polymerase behavior makes the de novo design of nucleotide analogs difficult. Moreover, it emphasizes the importance of examining nucleotide analogs in both the template and growing strands of DNA.

In addition to single nucleotide analogs, much effort has been focused on the search for novel base pairs that can function seamlessly in DNA (7). The addition of one or more new base pairs in DNA has great potential for applications as wide ranging as data storage, molecular computers and biochemical insight into the function of DNA polymerases themselves. However the discovery of a true third base pair remains elusive. A significant hindrance to progress in this area is the time necessary to synthesize and test potential novel base pairs. Thus far, research into nucleotide analogs has relied on one at a time trial and error approach. This results in a substantial amount of time being spent on failed base pairs. In order to circumvent this, we have developed a new approach to base pair discovery that allows for the selection of biologically acceptable base pairs from a pool of nucleotides (Figure 1). The key features of our approach are the use of a DNA polymerase as the selection tool and mass spectrometry as the identification tool. Mass spectrometry has been used previously to identify naturally modified nucleotides in RNA (9). Through this, it has been shown to be a capable method of analysis for identification of a single compound in a heterogeneous mixture. Furthermore, the sensitivity of LC-MS detection makes it a feasible tool for the analysis of very small quantities of material.
As a proof of principle, we first tested the selection scheme on a known unnatural base pair. For this, the bases 2,4-diaminopyrimidine (‘\(k\)’) and xanthine (‘X’) (Figure 2) were chosen for their known ability to be moderately incorporated by some DNA polymerases (10,11). Using a template containing \(d_k\) and the Klenow Fragment of *Escherichia coli* DNA polymerase I (KF), we were able to both selectively identify \(d_XTP\) from a pool of nucleotides and detect \(d_XTP\) at low levels of incorporation (\(\sim 10\%\)).

**Figure 1.** Overview of nucleotide selection scheme. A template containing a nucleotide analog is incubated with a library of nucleoside triphosphate analogs and a DNA polymerase. The extended oligonucleotides are separated, enzymatically digested to nucleosides and analyzed by LC-MS.

**Figure 2.** The positive control unnatural base pair: 2,4-diaminopyrimidine (\(k\)) and xanthine (X).

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**MATERIALS AND METHODS**

All reagents were purchased from Sigma-Aldrich unless otherwise noted. Enzymes were purchased from New England Biolabs, unless otherwise noted. Solvents were distilled immediately before use and synthetic reactions were carried out under \(N_2\) (g). The synthesis of \(d_k\) required significant alteration from the published procedure (Supplemental Data). \(d_XTP\) was synthesized as reported previously (12). DNA templates were synthesized on an ABI 394 DNA synthesizer.

**5’ End labeling and annealing of DNA**

DNA was \(5’-[\text{\(^{32}\)}P]P\)-labeled using polynucleotide kinase and \([\gamma-\text{\(^{32}\)}P]\)ATP, and the reaction was stopped by heating to 65°C for 15 min. Slow cooling of this mixture allowed for annealing of the self-priming template. DNA was desalted before use on a P2 BioGel column.

**Polymerization assays with pol \(\alpha\) and KF**

Assays contained 1 \(\mu\)M to 1 mM \(5’-[\text{\(^{32}\)}P]P\) DNA, 50 mM Tris–HCl (pH 7.6), 10 mM MgCl\(_2\), 1 mM DTT, 0.05 mg/ml BSA and various concentrations of \(d_XTP\) and enzyme. For the selection method, polymerization reactions were initiated by the addition of KF (0.5 U/\(\mu\)l), incubated at 37°C for 60 min and quenched by adding an equal volume of gel loading buffer (90% formamide in 1× TBE). Quenched reaction mixtures were loaded directly onto a polyacrylamide gel (20% acrylamide, 1× TBE, 0.8 mm × 300 mm × 375 mm).

**DNA recovery**

The primer and primer +1 bands in the polyacrylamide gel were visualized by autoradiography. After localization, the primer +1 band was carefully excised and the DNA was extracted. The extraction procedure used was a modification of published procedures (13). Briefly, the sample was thoroughly crushed in a small tube, then subjected to three consecutive extractions with 0.5 ml buffer each [5 M \(\text{NH}_4\text{Ac}_2\), 1 mM EDTA and 10 mM Mg(AcO)\(_2\)]. The extractions, all on a rotator, were of the following durations: (i) 3 h at 37°C; (ii) overnight at 4°C; (iii) 3 h at 37°C. After each extraction, the mixture was filtered and the filtrate saved; after the final extraction, the gel was rinsed with a further 0.5 ml buffer. The combined filtrates were then dried to a solid, which was desalted on a P2 Bio-Gel column.

**Oligonucleotide digestion**

DNA was incubated with calf intestinal phosphodiesterase I (Sigma, 7 \(\mu\)U per 1 nmol reaction; this preparation also contains alkaline phosphatase as a contaminant, and thus it gives nucleosides as the product) at 37°C for 2 h. Following this, the proteins were removed using a Microcon-10 filtration tube. The samples were washed twice to ensure complete recovery of the nucleosides. Nucleoside mixtures were desalted by flash chromatography using C18-derivitized silica. A column (60 mm × 5 mm) was pretreated with methanol and equilibrated with water. After a wash of one column volume water, the nucleosides were eluted with 100% methanol. The methanol was removed by evaporation before LC-MS analysis.
LC-MS
Liquid chromatography was performed on an Agilent C18 microbore column (50 μM diameter) at a flow rate of 10 μl/min. The LC eluent flowed directly into a PE Sciex Q-Star Pulsar mass spectrometer. Mass standards consisted of a solution of PEG-200 enriched in the lower mass region by passing the solution through a reverse phase cartridge. Because the longer chains are retained slightly on reverse phase media, this effectively skews the molecular weight standard solution to a lower mass region, which is more useful for nucleosides. This solution was concurrently introduced into the spectrometer using a t-joint fitted between the LC outlet and the MS inlet.

RESULTS AND DISCUSSION
The selection strategy outlined in Figure 1 was developed using a known unnatural base pair. A short, self-pairing oligonucleotide containing dκ was incubated with KF and a pool of dNTPs. This allows the polymerase to select for a novel base pair through covalent linkage of the unnatural dNTP (Y) to the same DNA strand as its partner. Any base pair thus selected will be de facto biologically relevant because of the method of its selection. PAGE was used to separate the positive hits by virtue of the difference in mobility of the original DNA strand and one that is extended by one additional nucleotide. Subsequent enzymatic digestion of the extracted oligonucleotides and mass spectral analysis of the resultant nucleosides then provides the molecular formula of the positive hits. From this, the identity of the incorporated nucleotide is easily determined.

Several DNA polymerases and template sequences were tested for use in the primer extension selection (Figure 3). In designing the DNA strand, we sought to use the shortest possible sequence for two reasons: (i) a smaller number of natural nucleosides in the final LC-MS sample increases the ability to detect the lower abundance unnatural nucleoside; and (ii) baseline separation of the primer and primer +1 bands by PAGE is necessary for accurate identification of only extended products, and is more easily achieved for smaller pieces of DNA. Additionally, the short loop sequence was designed to give maximum duplex stability (14,15). We found that the 23mer shown in Figure 4 provided an optimum primer:template for balancing a short sequence with enzymatic activity (Figure 4). In this sequence context, both KF and pol α incorporated dXTP into DNA across from template dκ (Figure 5). However, Moloney murine leukemia virus reverse transcriptase and T4 DNA polymerase did not detectably incorporate the analog (<1%; data not shown). As KF incorporates dXTP opposite dκ much more effectively than does pol α, all further studies were performed with this enzyme. Manipulation of concentrations of dXTP and KF allowed us to control the amount of primer extension, and thereby assess the threshold for nucleoside detection by this method.

After quenching the reactions and separating the primer and primer +1 products by PAGE, the products were located by autoradiography and the primer +1 product extracted from the gel. Optimization of extraction conditions resulted in a maximum recovery of ~70% of the DNA from the gel. This was then enzymatically digested to yield the free nucleosides. The crude mixture of nucleosides was passed through a C18 column before analysis by LC-MS in order to reduce the amount of salt, which contaminates the detector.

Initial studies identified LC conditions that gave efficient separation of nucleosides and MS parameters for the sensitive detection of the six nucleosides present in the mixture (Table 1). In order to obtain a highly accurate mass of the nucleoside samples, they must contain an internal mass standard. This was accomplished using a post-column t-joint to concurrently infuse both the sample and a solution of molecular weight standards into the MS. Detection and mass analysis

Figure 3. Sequences of oligonucleotides studied.

Figure 4. Testing the DNA sequences in Figure 3 for activity as substrates for pol α and KF. The polymerase used and the oligonucleotide length are indicated in the figure. For each length template, the 5 lanes are the: 1, no enzyme control; 2, no dNTP control; 3, +1 extension; 4, +2 extension; and 5, full extension.
parameters were then optimized using mixtures of purified nucleosides in similar concentrations to that of predicted samples from the selection scheme.

In order to fully test the scheme, we subjected 1 nmol of DNA containing dX (Figure 4, DNAx) to a round of selection with dXTP, under several different conditions. In one set of experiments, the template DNA was incubated with KF and a range of concentrations of dXTP to assess the lower limit of incorporation needed for detection of a positive hit. In another set, DNAx and KF were incubated with dXTP in the presence of a variety of other nucleotides, both natural and unnatural (Figure 6). This latter experiment served two purposes: (i) to test if the presence of inactive dNTPs will interfere with the detection of the active ones; and (ii) to verify that the nucleotides which should not be incorporated are not detected as false positives.

In all samples, dX was easily detected, even at <1% predicted extension of the primer (i.e. <10 pmol product). Additionally, the accuracy of the mass assigned was quite good, 3.2 mmu or less from the exact mass (most samples <1 mmu from the exact mass). This level of accuracy allows for the deduction of the molecular formula of the molecule detected, providing an unambiguous identification of the nucleoside. The results for the incoming nucleotide dXTP were also positive, though less straightforward than for dX. In the absence of other dNTPs, dX was detected (as its free base, xanthine) in all samples, save the one containing the lowest concentration of dXTP, implying that incorporation must be 10% to detect a hit (~100 pmol). When it was detected, its experimental mass was <4 mmu from the exact mass of xanthine. On average, the accuracy was slightly less than was seen for dX, but was still within the range needed for the assignment of the molecular formula of the base (Table 2). A detailed example of the data analysis is included in Supplementary Data.

In those samples containing dNTP mixtures during the polymerization reaction, xanthine was readily detected. Under these conditions, the amount of elongated product was reduced ~6-fold as compared to the identical assay (100 μM dXTP) that lacked any other natural or analog dNTPs. Thus, other dNTPs inhibit dXTP polymerization but are not incorporated in place of dXTP, as predicted by the selectivity of KF for polymerization of dXTP opposite dX. Furthermore, we were unable to detect incorporation of any of the other analog dNTPs included in the mixtures, thereby providing a direct demonstration of the utility of using a DNA polymerase to identify a novel base pair from a library of dNTP analogs.

The enhanced sensitivity towards dX relative to dX may have resulted from several causes. Primarily, in the LC-MS

Table 1. LC method for nucleoside separation

<table>
<thead>
<tr>
<th>Time (min)</th>
<th>% B</th>
</tr>
</thead>
<tbody>
<tr>
<td>0</td>
<td>0</td>
</tr>
<tr>
<td>5</td>
<td>0</td>
</tr>
<tr>
<td>20</td>
<td>15</td>
</tr>
<tr>
<td>22</td>
<td>50</td>
</tr>
<tr>
<td>23</td>
<td>50</td>
</tr>
<tr>
<td>25</td>
<td>0</td>
</tr>
</tbody>
</table>

Solvent A, 0.1% formic acid; Solvent B, 80% CH3CN in A.

Figure 5. Incorporation of dXTP opposite dX by KF and pol α. Polymerase and amount of dXTP used are indicated. The lengths of the DNA bands are indicated with arrows.

Table 2. Detection of the unnatural base pair by MS

<table>
<thead>
<tr>
<th>[dXTP] (μM)</th>
<th>Expected incorporation</th>
<th>Δmmu dX</th>
<th>Δmmu dX</th>
</tr>
</thead>
<tbody>
<tr>
<td>1</td>
<td>&lt;1% (&lt;10 pmol)</td>
<td>3.2</td>
<td>nd</td>
</tr>
<tr>
<td>10</td>
<td>10% (100 pmol)</td>
<td>0.2</td>
<td>0.5</td>
</tr>
<tr>
<td>50</td>
<td>25% (250 pmol)</td>
<td>0.3</td>
<td>3.9</td>
</tr>
<tr>
<td>100</td>
<td>55% (350 pmol)</td>
<td>0.6</td>
<td>0.2</td>
</tr>
<tr>
<td>500</td>
<td>&gt;50% (&gt;500 pmol)</td>
<td>0.7</td>
<td>1.0</td>
</tr>
<tr>
<td>MIXTURE A</td>
<td>&lt;35% (&lt;350 pmol)</td>
<td>1.5</td>
<td>0.7</td>
</tr>
<tr>
<td>MIXTURE B</td>
<td>&lt;35% (&lt;350 pmol)</td>
<td>0.1</td>
<td>2.5</td>
</tr>
<tr>
<td>MIXTURE C</td>
<td>&lt;35% (&lt;350 pmol)</td>
<td>0.7</td>
<td>0.6</td>
</tr>
</tbody>
</table>

dX was detected as the [MH]+ ion and dX as the [BH]+ ion (the free base form of dX). [dXTP] refers to the amount used in the first step of the selection scheme. The experiments were performed with 1 nmol of DNAx, and the expected amount of extended DNA was estimated from prior kinetic studies; for the mixtures, the incorporation of dXTP is expected to be significantly lower due to competition by the inactive dNTPs. Although the data were not quantified, visual inspection of the autoradiograms used to locate the products indicated that the amount of extension was similar to the predicted amount. Δmmu refers to the difference between the detected mass and the known mass in millimass units. nd, not detected.

MIXTURE A: dATP, dGTP, dCTP, TTP, dXTP

MIXTURE B: dXTP + the dNTPs shown below

MIXTURE C: MIXTURE A + MIXTURE B

Figure 6. Composition of the dNTP mixtures used in the full run-through of the selection scheme. The concentration of each nucleotide was 100 μM. dRTP, deoxyribose 5’triphosphate.
analysis of pure nucleoside solutions, dX reproducibly gave a less intense signal than the dx. This likely results from a difference in the case of protonation of the two molecular structures. Diaminopyrimidine has two free amino groups, which are relatively easy to protonate, whereas xanthine has none. The general trend of free amino groups leading to a more intense MS signal holds true for the four natural nucleosides as well. One additional possibility is that there was actually a higher concentration of dx due to contamination of unextended template in the samples. However, this seems unlikely because analytical scale reactions showed that overlap between the primer and primer +1 bands separated by gel electrophoresis is well under 1%.

A complicating factor with regard to the mass accuracy of the dX peak is contamination by other species. First, dX and deoxyguanosine slightly overlap during elution from the LC. Because the 13C isotope peak of [guanine-H]+ has the same mass as the protonated molecular peak of xanthosine, this leads to a slight skewing of the mass peak. Second, small amounts of non-nucleoside contaminants in the final sample are an inevitable result of the manipulations that the sample goes through before LC-MS analysis. Unfortunately, one of these contaminants gives a small peak that overlaps with xanthosine. This leads to a greater variability in the accuracy of the detection of dX. In the least concentrated sample (extension <1%), the contaminant completely obscures any dX that may be present. However, despite the above issues, the mass accuracy and sensitivity for detection of dX, though noticeably worse than for dx, is still quite reasonable and allows for ready identification of the incorporation of dXTP.

In looking at the broader implications of these results, this selection method clearly works. The incorporation of dXTP was readily detected at levels as low as 10% extension (100 pmol). Given that xanthine ionizes quite poorly, and that the template nucleoside, dx, was detected with even better mass accuracy and sensitivity, we feel that 10% incorporation represents a true lower limit of detection using this system. Importantly, incorporation of dx was detected in mixtures of several natural and unnatural nucleotides, and as predicted based on the studies using KF and individual analogs (data not shown), incorporation of no other nucleoside analogs was detected in these samples (i.e. no false positives). This gives direct proof that this method can be used to detect a novel base pair from a pool of nucleotide analogs.

We have demonstrated that our new selection strategy is a feasible one for use in the discovery of novel base pairs. This, coupled with the potential applications for the discovery of novel substrates for DNA polymerases using natural templates, makes this method a useful new tool for nucleotide analog research.

SUPPLEMENTARY DATA
Supplementary Data are available at NAR Online.

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